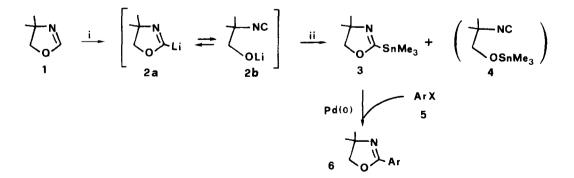
Masked Multifunctionalization of Aromatics by Palladium-catalyzed Halogen-Oxazoline Exchange

Alessandro Dondoni,^{*} Marco Fogagnolo, Giancarlo Fantin, Alessandro Medici, and Paola Pedrini Dipartimento di Chimica, Laboratorio di Chimica Organica, Università, Ferrara, Italy

<u>Abstract:</u> The $(Ph_3^P)_4$ Pd-catalyzed cross-coupling of aryl and heteroaryl halides with 2-(trime-thylstannyl)-4,4-dimethyl-2-oxazoline **3** affords aromatic 2-oxazolines **6** in very high yields.

The transition metal-catalyzed cross-coupling of aryl halides with organometallic species represents one of the most straightforward and versatile methods for carbon-carbon bond formation at aromatic rings.¹ The use of metalated heterocycles² in this methodology may provide a tool for aromatic functionalization in case the selected heterocycle is a masked functional group(s) equivalent. We report here the efficient palladium-catalyzed cross-coupling of a stannyloxazoline with aryl and heteroaryl halides to give the corresponding 2-oxazolines in very high yields. The synthetic utility of the 2-oxazolinyl group as a masked functionality (CHO, CO_H) and director of metalation as well as nucleophilic and electrophilic aromatic substitutions has been amply demonstrated. Common routes to aromatic oxazolines are based on the oxazoline ring closure at carboxylic acid derivatives; 3 a single method exists. limited in scope however, dealing with the cross-coupling of aryl Grignards and a 2-methylthiooxazoline. The hitherto unreported 2-(trimethylstannyl)-4,4-dimethyl-2-oxazoline 3 (bp 78-80°C at 16 mmHg) was prepared (70-80% yield) by lithiation (BuLi) of 4,4-dimethyl-2-oxazoline ${f 1}$ and then treatment with trimethyltin chloride (Scheme). Evidently almost exclusive quenching of lithio-oxazoline 2a by the alkyltin chloride takes place since the O-trimethylstannyl isocyanide **4** arising from the open-chain lithium alkoxide **2b** was detected only by IR (2000 cm⁻¹) in the crude reaction mixture. 5 Cross-coupling reactions of the stannyloxazoline **3** (2.2 mmol) with the selected organic halide 5 (2 mmol) were carried out in refluxing benzene (15 mL, 12-20 h)



Scheme. Reagents and Conditions: i, BuLi (1.1 equiv.), -78°C in Et 0; ii, Me SnCl (1.1 equiv.); iii, (Ph P) Pd (5% equiv.), C H, reflux.

under Argon atmosphere in the presence of 5% equiv. of tetrakis(triphenylphosphine)palladium(0). The resulting aryl and heteroaryl 2-oxazolines \vdots were isolated by column chromatography (silica, diethyl ether : <u>n</u>-hexane 1:1) (Table). The wide scope of the method is evident when considering that besides giving products in very high yields it applies equally well to a variety of organic halides differing for the nature of the aromatic ring as well as the electron-donor or acceptor character of the substituent. As expected from reactivities of organic halides towards organostannanes, ¹ aromatic iodides and bromides were quite efficient but a chloride was also successfully employed. Therefore this direct entry to aromatic oxazolines from organic halides nicely complements the Meyers methods ³ when the prerequisite carboxylic acids or nitriles are not available. The synthetic utility of this halogen-oxazoline exchange can be foreseen since it converts aryl and heteroaryl halides into masked carbonyl and carboxylic acid derivatives.

TABLE. Aromatic 2-oxazolines 6 from Cross-Coupling of Stannyloxazoline 3 and Organic Halides 5

ArX 5	2-0xazoline 6^a	t/h	Yield % ^b
PhI	Ar = Ph	12	100
MeOC_H_Br-p	p-MeOC_H	12	85
MeSC ⁶ ⁴ Br-p	p-MeSC H 6 4 p-FC H	15	100
FC_H_Br-p MeCOC_H_Br-p	p-FC _c H _c	24	100
MeCOC_H_Br-p	p - M = 0 p - M = COC H p - M = COC H q = M = 0 H	12	93
β -NaphBr ^c	β -Naph 6.4	12	95
2-BrPy _d	2-Py	24	85
J-BrPy	З-Ру	24	82
3-BrPye 2-ClQye	2-Qy	12	75
3-BrQy _	З-Qу	12	90
4-BrQy-i	4-Qy-i	12	92
3-BrQy 4-BrQy-i 3-BrTheng	3-Then	12	100
3-BrFur	3-Fur	18	85

^a All products were characterized by Mass and NMR spectra and showed identical properties to those of the literature. Isolated yields. β -Naph = β -naphtyl. Py = pyridyl. Qy = quinolyl. Qy-i = isoquinolyl. Then = thienyl. Fur = furyl.

References and Notes

- Reviews: M. Kumada <u>Pure Appl. Chem.</u>, 52,669 (1980); E. Negishi <u>Acc. Chem. Res.</u>, 15, 340 (1982). For selected papers and leading references see: I. Kuwajima, H. Urabe, <u>J. Am. Chem. Soc.</u>, 104, 831 (1982); F. Ozawa, N. Kawasaki, T. Yamamoto, A. Yamamoto, <u>Chem. Lett.</u>, 567 (1985); Y. Yokoyama; S. Ito, Y. Takahashi, Y. Murakami, <u>Tetrahedron Lett.</u>, 26,6457 (1985); T. Hayashi, M. Konishi, Y. Kobori, M. Kumada, T. Higuchi, <u>K. Hirotsu</u>, <u>J. Am. Chem. Soc.</u>, 106, 158 (1984).
- Y. Yamamoto, A. Yamagi, <u>Chem. Pharm. Bull.</u>, **30**, 1731 and 2003 (1982); A. Minato, K. Suzuki,
 K. Tamao, M. Kumada, J. <u>Chem. Soc. Chem. Commun.</u>, 511 (1984); A. Carpita, R. Rossi, <u>Gazzetta</u>,
 115, 575 (1985); A. Dondoni, M. Fogagnolo, A. Medici, E. Negrini, Synthesis, 1986 (in press).
- 3. A. I. Meyers, E. D. Mihelich in "New Synthetic Methods", Verlag Chemie, Wheinheim, vol. 5, p. 105 (1979); M. Reuman, A. I. Meyers, Tetrahedron, 41, 837 (1985).
- 4. L. N. Pridgen, L. B. Killmer, J. Org. Chem., 46, 5402 (1981).
- 5. By contrast, quenching the above equilibrium mixture with trimethylsilyl chloride gave exclusively <u>0</u>-trimethylsilyl β -hydroxyethylisocyanide. For a summary of the ambident chemistry of 2-lithio derivatives of oxazoles and 2-oxazolines see ref. 4 and A. Dondoni, T. Dall'Occo, G. Fantin, M. Fogagnolo, A. Medici, P. Pedrini, J. C. S. Chem. Commun., 258 (1984); A. Dondoni, Lect. in Heterocycl. Chem., 8, 13 (1985).

(Received in UK 14 August 1986)