## **Masked Multifunctionalization of Aromatics by Palladium-catalyzed Halogen-Oxazoline Exchange**

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Abstract: The (Ph<sub>2</sub>P)<sub>4</sub>Pd-catalyzed cross-coupling of aryl and heteroaryl halides with 2-(trimethylstannyl)-4,4-dimethyl-2-oxazoline 3 affords aromatic 2-oxazolines 6 in very high yields.

The transition metal-catalyzed cross-coupling of aryl halides with organometallic species represents one of the most straightforward and versatile methods for carbon-carbon bond formation at aromatic rings. The use of metalated heterocycles in this methodology may provide a tool for aromatic functionalization in case the selected heterocycle is a masked functional group(s) equivalent. We report here the efficient palladium-catalyzed cross-coupling of a stannyloxazoline with aryl and heteroaryl halides to give the corresponding 2-oxazolines in very high yields. The synthetic utility of the 2-oxazolinyl group as a masked functionality (CHO,  $CO<sub>4</sub>H$ ) and director of metalation as well as nucleophilic and electrophilic aromatic substitutions has been amply demonstrated. 3 Common routes to aromatic oxazolines are based on the oxazoline ring closure at carboxylic acid derivatives;  $^3$  a single method exists, limited in scope however, dealing with the cross-coupling of aryl Grignards and a 2-methylthiooxazoline. 4 The hitherto unreported 2-(trimethylstannyl)-4,4-dimethyl-2-oxazoline 3 (bp 78-80°C at 16 mmHg) was prepared (70-80% yield) by lithiation (BuLi) of 4,4-dimethyl-2-oxazoline **1** and then treatment with trimethyltin chloride (Scheme). Evidently almost exclusive quenching of lithio-oxazoline **2a** by the alkyltin chloride takes place since the g-trimethylstannyl isocyanide -1 4 arising from the open-chain lithium alkoxide **2b** was detected only by IR (2000 cm ) in the crude reaction mixture.  $^5$  Cross-coupling reactions of the stannyloxazoline  $\,3\,$  (2.2 mmol) with the selected organic halide 5 (2 mmol) were carried out in refluxing benzene (15 mL, 12-20 h)



Scheme. Reagents and Conditions: i, BuLi (1.1 equiv.), -78°C in Et<sub>2</sub>0; ii, Me<sub>3</sub>SnCl (1.1 equiv.); iii,  $\left(\mathsf{Ph}_{3}P\right)_{4}Pd$  (5% equiv.),  $C_{6}H_{6}$ , reflux.

under Argon atmosphere in the presence of 5% equiv. of tetrakis(triphenylphosphine)palladium(0). The resulting aryl and heteroaryl 2-oxazolines U were isolated by column chromatography (silica, diethyl ether : n-hexane 1:1) (Table). The wide scope of the method is evident when considering that besides giving products in very high yields it applies equally well to a variety of organic halides differing for the nature of the aromatic ring as well as the electron-donor or acceptor character of the substituent. As expected from reactivities of organic halides towards organostannanes, aromatic iodides and bromides were quite efficient but a chloride was also successfully employed. Therefore this direct entry to aromatic oxazolines from organic halides nicely complements the Mevers methods<sup>3</sup> when the prerequisite carboxylic acids or nitriles are not available. The synthetic utility of this halogen-oxazoline exchange can be foreseen since it converts aryl and heteroaryl halides into masked carbonyl and carboxylic acid derivatives.

TABLE. Aromatic 2-oxazolines 6 from Cross-Coupling of Stannyloxazoline 3 and Organic Halides 5



<sup>a</sup> All products were characterized by Mass and NMR spectra and showed identical properties to those of the literature. Isolated yields.  $\beta_{\overline{n}}$ Naph = β-naphtyl. <sup>d</sup> Py = pyridyl. <sup>e</sup> Qy = quinolyl.  $\begin{bmatrix} 1 & 0 & 0 \\ 0 & 1 & 1 \end{bmatrix}$  = isoquinolyl.  $\begin{bmatrix} 2 & 0 & 0 \\ 0 & 1 & 1 \end{bmatrix}$  Then = thienyl.  $^{\prime}$  Fur = furvl.

## References and Notes

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- 5. By contrast, quenching the above equilibrium mixture with trimethylsilyl chloride gave exclusively  $Q$ -trimethylsilyl  $\beta$ -hydroxyethylisocyanide. For a summary of the ambident chemistry of 2-lithio derivatives of oxazoles and 2-oxazolines see ref. 4 and A. Dondoni, T. Dall'Occo, G. Fantin, M. Fogagnolo, A. Medici, P. Pedrini, J. C. S. Chem. Commun., 258 (1984); A. Dondoni, Lect. in Heterocycl. Chem., 8, 13 (1985).

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